## **Supplementary Information**

General Procedures:

General procedure 1. Cycloaddition of nitrone (–)-1 and allyl glycosides 2:

Allyl glycosides **2** (1 mmol) and nitrone **1** (1.2 mmol) are dissolved in 5 ml toluene and refluxed for appr. 3 days. After completeness (checked by TLC) of the reaction the solvent is reduced under reduced pressure and the remaining product is purified by column chromatography (silica gel, petrol ether/ ethyl acetate = 1:1) and can be isolated as a colourless oil.

General procedure 2. Sm<sup>II</sup>-mediated cleavage of isoxazolidines 3:

Preparation of a fresh solution of  $SmI_2$ : To Sm powder (150 mg, 1 mmol) in 2 ml THF under argon atmosphere  $C_2H_4I_2$  (160 mg, ) is added and stirred at room temperature, until all Sm has been oxidized. The solutions becomes dark blue. (We always used freshly prepared  $SmI_2$  solutions)

To this solution isoxazolidine **3** (0.2 mmol, dissolved in 3 ml THF and 0.5 ml abs. methanol) is added and stirred at room temperature until the colour of the solution has turned from blue to yellow. Subsequently, NH<sub>4</sub>Cl (sat.) is added and the reaction mixture extracted with ethyl acetate. The combined organic phases are dried over MgSO<sub>4</sub>, the solvent is evaporated and the remaining oily residue is purified by chromatography (silica gel, petrol ether/ ethyl acetate = 1:1). The product can be isolated as a colourless oil.

General procedure 3. Cleavage of the chiral auxiliary and of the amide in 4:

N,N-acetal **4** (0.12 mmol) is dissolved in 2N HCl (12 ml) and acetic acid (10 ml) and the reaction mixture is heated up to 80 °C for 2 h. After cleavage of the chiral auxiliary, the reaction mixture is evaporated to dryness. The remaining residue is dissolved in THF (4 ml) and  $H_2O$  (4 ml) and treted with LiOH (80 mg, 3.3 mmol). After appr. 2 h (TLC-control) the cleavage of the amide bond is completed and the residue is again evaporated to dryness. After  $CH_2Cl_2$  (10 ml) has been added and neutralization ( $H_2O$ , 2x5ml), the amont of solvent is diminished by evaporation. Subsequently, by addition of petrol ether the product **5** can be isolated as a colourless solid in quantitative yield.

α-**3b**:  $R_f = 0.56$  (silica, petrol ether/EtOAc = 1/1);  $[\alpha]_D^{20} = +58.9^\circ$  (c = 0.87, CH<sub>2</sub>Cl<sub>2</sub>); IR:  $\tilde{v}_{max} = 3022, 2956, 2919, 2862, 1698, 1450, 1382, 1093; <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>, 25 °C): δ =$ 

7.46 – 7.11 (m, 20H, CH-Benzyl), 4.99 – 4.49 (m, 8H, CH<sub>2</sub>-Benzyl), 4.40 –4.29 (m, 1H, CH), 3.99 –4.03 (m, 2H, 2xCH), 3.62 – 3.80 (m, 6H, 4xCH, CH<sub>2</sub>), 2.63 (s, 4H, NCH<sub>3</sub>, *H*CH), 2,29 – 1.19 (m, 12H, 3xCH, 5xCH<sub>2</sub>), 1.02 (d,  ${}^{3}J$  = 4.9 Hz, 3H, CH<sub>3</sub>), 0.91 (d,  ${}^{3}J$  = 6.5 Hz, 6H, 2xCH<sub>3</sub>);  ${}^{13}$ C-NMR (50 MHz, CDCl<sub>3</sub>, 25 °C, TMS):  $\delta$  = 173.40 (C=O), 139.14 (q), 138.66 (q), 138.52 (q), 138.47 (q), 128.02 – 128.81 (CH-Benzyl), 89.95 (q), 82.60 (CH), 79.78 (CH), 78.39 (CH), 75.83 (CH<sub>2</sub>-Benzyl), 75.41 (CH<sub>2</sub>-Benzyl), 73.96 (CH<sub>2</sub>-Benzyl), 73.44 (CH<sub>2</sub>), 73.06 (CH-Benzyl), 72.07 (CH<sub>2</sub>), 71.64 (CH), 69.45 (CH<sub>2</sub>), 66.51 (CH), 48.60 (CH), 40.85 (CH<sub>2</sub>), 39.20 (CH<sub>2</sub>), 35.18 (CH<sub>2</sub>), 30.31 (CH), 28.41 (CH<sub>2</sub>), 26.36 (NCH<sub>3</sub>), 24.82 (CH), 24.62 (CH<sub>3</sub>), 22.94 (CH<sub>2</sub>), 22.85 (CH<sub>3</sub>), 18.92 (CH<sub>3</sub>); elemental analysis: cal. C<sub>50</sub>H<sub>62</sub>N<sub>2</sub>O<sub>7</sub>: C 74.48, H 7.78, N 3.49; found: C 74.70, H 7.76, N 3.80.

 $\alpha$ -**4b**: R<sub>f</sub> = 0.23 (silica, petrol ether/EtOAc = 1/1);  $\left[\alpha\right]_D^{20}$  = +43.4° (c = 0.94, CH<sub>2</sub>Cl<sub>2</sub>); IR:  $\tilde{v}_{\text{max}}$  = 3337, 3027, 2955, 2867, 1698,1455, 1093; <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>, 25 °C, TMS): δ = 7.45 – 7.15 (m, 20H, CH-Benzyl), 5.03 – 4.48 (m, 8H, CH<sub>2</sub>-Benzyl), 4.11 – 4.00 (m, 1H, CH), 3.85 –3.67 (m, 8H, 6xCH, CH<sub>2</sub>), 2.77 (s, 3H, NCH<sub>3</sub>), 1.98 – 1.25 (m, 13H, 5xCH<sub>2</sub>, 3xCH), 1.07 – 0.87 (m, 9H, 3xCH<sub>3</sub>); <sup>13</sup>C-NMR (50 MHz, CDCl<sub>3</sub>, 25 °C): δ = 175.01 (C=O), 139.14 (q), 138.63 (q), 138.53 (q), 138.43 (q), 128.81 – 128.03 (CH-Benzyl), 82.61 (CH), 81.61 (q), 79.98 (CH), 78.47 (CH), 75.77 (CH<sub>2</sub>-Benzyl), 75.37 (CH<sub>2</sub>-Benzyl), 73.88 (CH<sub>2</sub>-Benzyl), 73.20 (CH<sub>2</sub>-Benzyl), 72.05 (CH), 71.07 (CH), 69.49 (CH<sub>2</sub>, C-6), 66.69 (CH<sub>2</sub>), 58.69 (CH), 48.44 (CH<sub>2</sub>), 47.15 (CH), 41.09 (CH<sub>2</sub>), 34.95 (CH<sub>2</sub>), 33,19 (CH<sub>2</sub>), 29.30 (CH), 25.79 (NCH<sub>3</sub>), 25.06 (CH), 24.38 (CH<sub>3</sub>), 22.68 (CH<sub>3</sub>), 22.60 (CH<sub>2</sub>), 18.89 (CH<sub>3</sub>); elemental analysis: cal. C<sub>50</sub>H<sub>64</sub>N<sub>2</sub>O<sub>7</sub>: C 74.60, H 8.01, N 3.48; found: C 74.20, H 7.67, N 3.41.

 $\alpha$ -**5b**: R<sub>f</sub> = 0.16 (silica, CH<sub>2</sub>Cl<sub>2</sub>/MeOH = 98/2); [α]<sub>D</sub><sup>20</sup> = +17.2° (c = 1.16, CH<sub>2</sub>Cl<sub>2</sub>); IR:  $\tilde{v}_{max}$  = 3379, 3011, 2929, 2846, 1734, 1650, 1100, 1036; <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>, 25 °C, TMS): δ = 7.49 – 7.34 (m, 20H, CH-Benzyl), 4.65 – 4.46 (CH<sub>2</sub>-Benzyl), 3.94 – 3.58 (m, 9H, 7XCH, CH<sub>2</sub>), 2.13 – 2.01 (m, 2H, CH<sub>2</sub>), 1.89 – 1.60 (m, 2H, CH<sub>2</sub>); <sup>13</sup>C-NMR (50 MHz, CDCl<sub>3</sub>, 25 °C): δ = 178.76 (C=O), 141.32 (q), 140.90 (q), 14.82 (q), 140.48 (q), 131.27 – 130.81 (CH-Benzyl), 84.54 (CH), 82.09 (CH), 80.69 (CH), 78.13 (CH<sub>2</sub>-Benzyl), 77.77 (CH<sub>2</sub>-Benzyl), 76.37 (CH<sub>2</sub>-Benzyl), 75.65 (CH<sub>2</sub>-Benzyl), 74.46 (CH), 73.15 (CH), 71.91 (CH<sub>2</sub>), 67.03 (CH), 54.45 (CH), 38.00 (CH<sub>2</sub>), 36.27 (CH<sub>2</sub>); elemental analysis: cal. C<sub>39</sub>H<sub>45</sub>NO<sub>8</sub>: C 71.43, H 6.92, N 2.14; found: C 71.17, H 6.95, N 2.72.

## X-ray structure of **12**:

prepared by general procedure 1

 $C_{23}H_{31}BrN_2O_4$ ,  $M_r = 479.4$ , orthorhombic, space group P  $2_12_12$  (No 18), T = 203(2) K, a = 13.052(3), b = 18.329(5), c = 9.700(9) Å, V = 2321(3) ų, Z = 4,  $D_c = 1.372$  g/cm³,  $\mu$ (MoK $\alpha$ ) = 1.802 mm⁻¹, F(000) = 1000, crystal size 0.60 x 0.50 x 0.10 mm³; Bruker AXS P4 diffractometer, MoK $\alpha$  radiation, graphite monochromator,  $\omega$ -scan,  $2.6 < \Theta < 25^\circ$ , h: 0/-14, k: 0/-20, 1: 1/-11, 1938 reflections collected, absorption correction via psi-scans, min/max transmission 0.357/0.773. Structure solved by Direct and conventional Fourier methods, full-matrix least-squares refinement based on  $F^2$  and 272 parameters, all but H-atoms refined anisotropically, H-atoms ,riding 'at idealized positions, refinement converged at R1(I >  $2\sigma$ (I)) = 0.079, wR2(all data) = 0.175, S = 1.055, absolute structure parameter refined to 0.00(3), min/max height in final  $\Delta F$  map -0.30/0.46 e/ų. Programs used: SHELXTL NT V5.10.

Figure 2. Molecular structure of **12.** Displacement ellopsoids shown at the 50% level, Hatoms omitted for clarity.

